# organic compounds

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# 4-Ethynyl-1,2-methylenedioxybenzene at 150 K

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The title compound,  $C_9H_6O_2$ , contains two moderate  $C-H\cdots O$  hydrogen bonds. That involving the terminal alkyne gives rise to chains along the *b* axis. The other hydrogen bond occurs over a centre of symmetry, leading to dimers. The combination of the two interactions gives rise to rings, each comprising six molecules, which are part of infinite sheets in the *bc* plane.

## Comment

The packing of simple aromatic compounds has attracted interest (Desiraju & Kishan, 1989; Weiss *et al.*, 1997). It has been demonstrated by Sarma & Desiraju (1986) that the effects of heteroatoms (principally O or Cl) either within the aromatic ring or peripheral to it can be structure determining. A low number of heteroatoms means that structures tend to adopt herring-bone motifs; benzene, with none, is the archetypal example (Bacon *et al.*, 1964). On the other hand, higher numbers of heteroatoms can lead to stacked motifs, *e.g.* pyranthrone (Sarma & Desiraju, 1986). Desiraju & Kishan (1989) demonstrated this principle with structural studies on 3,4-(methylenedioxy)cinnamic acid and [3,4-(methylenedioxy)phenyl]propiolic acid. Therefore, we decided to examine the structural motifs present in the related compound 4-ethynyl-1,2-methylenedioxybenzene, (I).



Compound (I) lies in a general position (Fig. 1), but with the non-H atoms coplanar to within 0.008 Å. All molecular geometry parameters lie within the normal ranges (Allen *et al.*, 1987).

The most significant feature of (I) is the presence of two intermolecular C-H···O hydrogen bonds (Table 1). The first occurs over a centre of symmetry so that the aromatic H2 atom acts as a donor to O9<sup>ii</sup> [symmetry code: (ii) 1 - x, 1 - y,

-z], generating a dimeric centrosymmetric  $R_2^2(8)$  motif (Fig. 2).

The second intermolecular hydrogen bond involves the alkynyl H8 atom, which acts as a donor to *n*-glide-related O9<sup>i</sup> [symmetry code: (i)  $\frac{3}{2} - x$ ,  $\frac{1}{2} + y$ ,  $\frac{1}{2} - z$ ]. This generates *C*(7) chains along the *y* axis, leading to herring-bone chains in the *ab* plane. The top half of each dimer is related to the bottom half of the neighbouring dimer by this C-H···O interaction (Fig. 2). Therefore, the dimers are related to each other in a zigzag manner along the *b* direction within the *bc* plane. The



#### Figure 1

A view of the structure of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

combination of the two hydrogen bonds generates a six-molecule  $R_6^4(28)$  ring as part of an infinite two-dimensional sheet (Fig. 2). There is also a possible C5-H5···O11<sup>iii</sup> interaction [symmetry code: (iii) -x, 1 - y, 1 - z] between dimers related by a centre of symmetry in the middle of the  $R_6^4(28)$  ring.



#### Figure 2

The packing diagram of (I) showing part of an infinite two-dimensional sheet in the *bc* plane. The component dimeric units and six-molecule rings are visible [symmetry codes: (i)  $\frac{3}{2} - x$ ,  $\frac{1}{2} + y$ ,  $\frac{1}{2} - z$ ; (ii) 1 - x, 1 - y, -z].

# **Experimental**

Compound (I) (Pelter *et al.*, 1990) was prepared according to the method of Corey & Fuchs (1972). 3,4-(Methylenedioxy)benzaldehyde was firstly transformed to the dibromoalkene by treatment with carbon tetrabromide, triphenylphosphine and zinc; subsequent treatment with butyllithium generated the title alkyne, (I), which exists as a supercooled oil at room temperature. The sample was placed in a sealed capillary and cooled *in situ* on the diffractometer using an Oxford Cryosystems open-flow cryostat (Cosier & Glazer, 1986). The crystal was grown by repeated cycling of the temperature across the freezing point of the compound (260 K). Crystal data

$C_0H_6O_2$
$M_r = 146.14$
Monoclinic, $P2_1/n$
a = 6.318 (3)  Å
b = 15.867 (10)  Å
c = 7.571 (4) Å
$\beta = 115.03 (3)^{\circ}$
$V = 687.7 (7) \text{ Å}^3$
Z = 4

# Data collection

## Refinement

Refinement on $F^2$	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.086$	$w = 1/[\sigma^2(F_o^2) + 1.6470P]$
$wR(F^2) = 0.163$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.153	$(\Delta/\sigma)_{\rm max} < 0.001$
1356 reflections	$\Delta \rho_{\rm max} = 0.31 \text{ e } \text{\AA}^{-3}$
100 parameters	$\Delta \rho_{\rm min} = -0.35 \ {\rm e} \ {\rm \AA}^{-3}$

 $D_x = 1.411 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation

reflections

 $\mu=0.100~\mathrm{mm}^{-1}$ 

 $\theta = 10.2\text{--}12.6^{\circ}$ 

T = 150 (2) KNeedle, pale yellow

 $\theta_{\rm max} = 26.04^{\circ}$  $h = -7 \rightarrow 7$ 

 $l = -9 \rightarrow 9$ 3 standard reflections

 $k = -19 \rightarrow 19$ 

Cell parameters from 40

 $0.19 \times 0.11 \times 0.08 \text{ mm}$ 

frequency: 60 min intensity decay: 9.6%

## Table 1

Geometry of hydrogen bonds and short intermolecular contacts (Å, °).

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$C8-H8\cdots O9^{i}$ $C2-H2\cdots O9^{ii}$ $C5-H5\cdots O11^{iii}$	0.95	2.58	3.526 (7)	178
	0.95	2.53	3.344 (7)	143
	0.95	2.70	3.529 (7)	146

Symmetry codes: (i)  $\frac{3}{2} - x$ ,  $\frac{1}{2} + y$ ,  $\frac{1}{2} - z$ ; (ii) 1 - x, 1 - y, -z; (iii) -x, 1 - y, 1 - z.

All H atoms were clearly defined in difference maps and were then treated as riding, with C-H = 0.95 or 0.99 Å. Examination of the structure with *PLATON* (Spek, 2000) showed that there were no

solvent-accessible voids in the crystal lattice. The high value of  $R_{\rm int}$  is attributed to the presence, within the region of the capillary irradiated by the X-ray beam, of additional crystallites of (I), the presence of which is an unavoidable consequence of the crystal-growth method.

Data collection: *STADI*4 (Stoe & Cie, 1997); cell refinement: *STADI*4; data reduction: *X-RED* (Stoe & Cie, 1997); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2000); software used to prepare material for publication: *SHELXL*97 and *PLATON*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: GD1130). Services for accessing these data are described at the back of the journal.

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